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(a) A bond formed by a shared pair of  
electrons with both electrons coming  
from the same atom. (b) A bond formed  
by a shared pair of electrons, each  
bonding atom contributing one electron

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to the pair. Answer: (i) (d) (ii) (a) Coordinate bond. (b) Covalent bond.

Question 19: (i) Acids dissolve in water to produce positively charged ions. Draw the structure of these positive ions.

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2• Define electrovalent bond. Solution: The chemical bond formed due to the electrostatic force of attraction between cation and anion is called an electrovalent bond. 3. Elements are classified as metals, non-metals, metalloids and inert gases. Which of

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them form electrovalent bond? Solution:

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students understand the concept very well.

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2) Covalent Bond: The chemical bond formed between two combining atoms by mutual sharing of one or more pairs of electrons. Compound formed due to this bond is called covalent compound.

3) Coordinate bond is the bond formed between two atoms by sharing a pair of electrons, provided entirely by one of the combining atoms but shared by both.

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The chemical bond formed due to the electrostatic force of attraction between a cation and an anion is called an electrovalent bond. The chemical bond formed between two combining atoms by mutual sharing of one or more pairs of electrons is called a covalent bond .

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Chemical Bonding and Molecular Structure Class 11 Notes Chemistry Chapter 4 • Chemical Bond The force that holds different atoms in a molecule is called chemical bond. • Octet Rule Atoms of different elements take part in chemical combination in order to complete their octet or to attain the noble gas configuration. • Valence Electrons

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Electrolysis is a process of decomposition of an ionic compound by passing electricity through it. Carbon and its Compounds – Bonding in Carbon Covalent compounds have low melting and boiling points.

## **ICSE Class 10th Chemistry 2 - Chemical Bonding - Ionic and ...**

The chemical compounds formed as a result of the transfer of electrons



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between metallic and non metallic atoms is called electrovalent/ionic bond. (c) A covalent compound. The molecule formed due to sharing of electrons between two nonmetallic atoms called a covalent compound.

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